

Math A, B

Math A: 1 ~ 17

Math B: 1 ~ 24, except 5

1. Numbers and expressions

a. Exponent rules

b. Factorization / expansion

$$\textcircled{1} \quad (a+b)(a^2 - ab + b^2) = a^3 + b^3$$

c. Real numbers

d. First-degree inequality

e. Absolute values

$$\textcircled{1} \quad \text{When } |x| = c, x = \pm c$$

2. Sets

a. Sets

$$\textcircled{1} \quad A \cap B, A \cup B, A \subset B, x \in A$$

b. Sufficient and necessary conditions

3. Quadratic equations

a. Quadratic functions

$$\textcircled{1} \quad y = ax^2 + bx + c$$

$$\textcircled{2} \quad y = a\left(x + \frac{b}{2a}\right)^2 - \frac{b^2 - 4ac}{4a}$$

$$\textcircled{3} \quad D = b^2 - 4ac$$

b. Minima, maxima

4. Figures and measurements

a. Sin, cos, tan

$$\textcircled{1} \quad 1 + \tan^2 \theta = \frac{1}{\cos^2 \theta}$$

$$\textcircled{2} \quad \sin(90^\circ - \theta) = \cos \theta$$

b. Trigonometric ratios

$$\textcircled{1} \quad \frac{a}{\sin A} = \frac{b}{\sin B} = \frac{c}{\sin C} = 2R$$

$$\textcircled{2} \quad a^2 = b^2 + c^2 - 2bc \cos A$$

5. Data analysis

a. Deviation

$$\textcircled{1} \quad s^2 = \bar{x}^2 - (\bar{x})^2$$

b. Tendency

6. Number of possible outcomes

a. Sets

$$\textcircled{1} \quad n(A \cup B) = n(A) + n(B) - n(A \cap B)$$

b. Permutation

$$\textcircled{1} \quad nP_r = \frac{n!}{(n-r)!} \quad (\text{Usually written as } P(n, r) \text{ outside Japan})$$

$$\textcircled{2} \quad \frac{(n-1)!}{2} \text{ for circular permutation}$$

c. Combination

$$\textcircled{1} \quad nC_r = \frac{nP_r}{r!} = \frac{n!}{r!(n-r)!} \quad (\text{Usually written as } \binom{n}{r} \text{ outside Japan})$$

$$\textcircled{2} \quad nC_r = nC_{n-r}$$

7. Probability

a. Properties of probability

$$\textcircled{1} \quad P(A) = \frac{n(A)}{n(U)}$$

$$\textcircled{2} \quad P(A) + P(\bar{A}) = 1$$

$$\textcircled{3} \quad P(A \cup B) = P(A) + P(B) - P(A \cap B)$$

b. Independent trials

$$\textcircled{1} \quad nC_r p^r (1-p)^{n-r}$$

c. Probability with constraints

$$\textcircled{1} \quad P_A(B) = \frac{P(A \cap B)}{P(A)}$$

8. Properties of figures

a. Ceva's theorem

$\textcircled{1}$ For ΔABC with points R, P, and Q on AB, BC, and CA

$$\text{respectively, } \frac{AR}{RB} \cdot \frac{BP}{PC} \cdot \frac{CQ}{QA} = 1$$

b. Menelaus's theorem

c. Power of a point theorem

9. Properties of integers

a. Number of divisors

$$\textcircled{1} \quad \text{If a natural number } N = p^a \cdot$$

$q^b \cdot r^c \cdots$, then the number of positive divisors = $(a+1)(b+1)(c+1)\cdots$ and the sum of all positive divisors = $(1+p+\cdots p^a)(1+q+\cdots q^b)(1+r+\cdots r^c)\cdots$

b. GCD, LCM

- ① If $a = ga'$ and $b = gb'$, then a' and b' are relative primes of each other, $l = ga'b'$, and $ab = gl$

c. Remainder rules

d. Euclidean algorithm

- ① If $ax + by = c$ and $ap + bq = c$, then $x = bk + p$ and $y = -ak + q$

e. Recurring decimal

f. Positional notation (base 2, 3, 4, ...)

10. Miscellaneous expressions

a. Binomial theorem

$$\textcircled{1} \quad \frac{n!}{p!q!r!} a^p b^q c^r$$

b. Arithmetic mean

- ① If $a > 0$ and $b > 0$, then $a + b \geq 2\sqrt{ab}$ is true and $a + b = 2\sqrt{ab}$ when $a = b$

c. Roots

$$\textcircled{1} \quad \sqrt{a+b \pm 2\sqrt{ab}} = \sqrt{a} \pm \sqrt{b}$$

d. Properties of complex numbers

$$\textcircled{1} \quad i^2 = -1$$

e. Solutions of functions

- ① If α and β are the two solutions of $ax^2 + bx + c = 0$, then $\alpha + \beta = -\frac{b}{a}$ and $\alpha\beta = \frac{c}{a}$

- ② If α , β , and γ are the three solutions of $ax^3 + bx^2 + cx + d = 0$, then $\alpha + \beta + \gamma = -\frac{b}{a}$ and

$$\alpha\beta + \beta\gamma + \gamma\alpha = \frac{c}{a}, \text{ and } \alpha\beta\gamma = -\frac{d}{a}$$

11. Figures and equations

a. Triangles

- ① For ΔABO with $O(0,0)$, $A(x_1, y_1)$, and $B(x_2, y_2)$, $S = \frac{1}{2}|x_1y_2 - x_2y_1|$

b. Circles

- ① $x^2 + y^2 = r^2$
- ② For a circle C $x^2 + y^2 = r^2$, the equation of the tangent line is $x_1x + y_1y = r^2$ for $P(x_1, y_1)$ on C

c. Optimization problems

12. Trigonometry

a. Trigonometric formulas

- ① $\sin(\alpha + \beta) = \sin \alpha \cos \beta + \sin \beta \cos \alpha,$
- $\cos(\alpha + \beta) = \cos \alpha \cos \beta - \sin \alpha \sin \beta,$
- $\tan(\alpha + \beta) = \frac{\tan \alpha + \tan \beta}{1 - \tan \alpha \tan \beta}$, etc.

$$\textcircled{2} \quad \sin^2 \theta = \frac{1-c}{2}, \quad \cos^2 \theta =$$

$$\frac{1+c}{2}, \quad 2 \sin \theta \cos \theta = \sin 2\theta$$

$$\textcircled{3} \quad \sin \alpha \cos \beta = \frac{1}{2}\{\sin(\alpha + \beta) + \sin(\alpha - \beta)\}, \text{ etc.}$$

$$\textcircled{4} \quad \sin A + \cos B = 2 \sin \frac{A+B}{2} \cos \frac{A-B}{2}, \text{ etc.}$$

$$\textcircled{5} \quad \sin 3\theta = 3 \sin \theta - 4 \sin^3 \theta, \\ \cos 3\theta = -3 \cos \theta + 4 \cos^3 \theta$$

⑥ Note: sec, csc, and cot are not included in the syllabus

13. Exponential and logarithmic function

a. Exponential and logarithmic rules

- ① Note: In Japan, $\log(x) = \log$

- base e ($\ln(x)$) and $\log_{10}(x) = \log$ base 10
- ③ $pa_{n+2} + qa_{n+1} + ra_n = 0$
14. Differential calculus 1
- Limit definitions
 - Basic differentiation
 - Optimization problems
15. Integral calculus 1
- Basic integration

① $\int_{\alpha}^{\beta} a(x - \alpha)(x - \beta)dx = -\frac{a}{6}(\beta - \alpha)^3$

② Note: Integral calculus 1 does not contain integration by parts
16. Vectors
- Dot product (cross product is not included)
 - Vector geometry (2D and 3D)

① If $a\vec{PA} + b\vec{PB} + c\vec{PC} = 0$ for a ΔABC and a point P, $\vec{AP} = k\left(\frac{n\vec{AB} + m\vec{AC}}{m+n}\right)$

② For \vec{p} on \overrightarrow{AB} , $\vec{p} = (1-t)\vec{a} + t\vec{b}$

③ For \vec{h} on ΔABC , $\vec{h} = s\vec{OA} + t\vec{OB} + u\vec{OC}$ ($s + t + u = 1$)
17. Sequences
- Arithmetic and geometric sequences
 - Sum of sequences

① $S_n = \frac{1}{2}n(a + l)$, $S_n = \frac{1}{2}n(2a + (n-1)d)$, $S_n = a\frac{r^{n-1}-1}{r-1} = a\frac{1-r^n}{1-r}$

② $\sum_{k=1}^n k = \frac{1}{2}n(n+1)$, $\sum_{k=1}^n k^2 = \frac{1}{6}n(n+1)(2n+1)$, $\sum_{k=1}^n k^3 = \left\{\frac{1}{2}n(n+1)\right\}^2$, etc.
 - Recurrence

① $a_{n+1} = a_n + d$

② $a_{n+1} = n(a_n)^m$
18. Complex plane
- De Moivre's theorem

① $(\cos \theta + i \sin \theta)^n = \cos n\theta + i \sin n\theta$
 - Nth root of 1

① $z_k = \cos \frac{2k\pi}{n} + i \sin \frac{2k\pi}{n}$
19. Curves on a plane
- Curves

① $y^2 = 4px$

② $\frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$ with $F(\sqrt{a^2 - b^2}, 0), F'(-\sqrt{a^2 - b^2}, 0)$

③ $\frac{x^2}{a^2} - \frac{y^2}{b^2} = 1$ with $F(\sqrt{a^2 + b^2}, 0), F'(-\sqrt{a^2 + b^2}, 0)$
 - Tangent lines
 - Parametric curves

① $\begin{cases} x = a \sin \theta \\ y = b \cos \theta \end{cases}$ etc.
 - Polar coordinates

① (r, θ)
20. Limits
- Limits of r^n

① Ex. $\lim_{n \rightarrow \infty} \frac{r^{n-1}}{r^{n+1}} = -1$ when $|r| < 1$, $= 0$ when $r = 1$, and $= 1$ when $|r| > 1$
 - Limits of sequences

① Ex. $a_{n+1} = \frac{2a_n - 3}{a_n - 4}$, $\lim_{n \rightarrow \infty} a_n$

② Ex. $\frac{1}{\sqrt{1+\sqrt{3}}} + \frac{1}{\sqrt{3+\sqrt{5}}} + \dots + \frac{1}{\sqrt{2n-1+\sqrt{2n+1}}} + \dots$

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③ Ex. Fractals

c. Trigonometric limits

$$\textcircled{1} \lim_{x \rightarrow 0} \frac{\sin x}{x} = 1$$

21. Differential calculus 2

a. Chain rule

b. Limits of e

$$\textcircled{1} \lim_{h \rightarrow 0} (1+h)^{\frac{1}{h}} = e$$

c. Implicit differentiation

d. Mean-value theorem

e. L'Hospital's rule

f. Optimization problems

g. Differentiation of parametric functions

h. Approximation

$$\textcircled{1} \text{ If } h \approx 0, \text{ then } f(a+h) \approx f(a) + f'(a)h$$

$$\textcircled{2} \text{ If } x \approx 0, \text{ then } f(x) \approx f(0) + f'(0)x$$

22. Integral calculus 2

a. Integration by parts

b. Trigonometric integration

c. Volume

d. Integration of parametric functions

e. Length of a curve

$$\textcircled{1} L = \int_a^b \sqrt{1+y'^2}$$

$$\textcircled{2} L = \int_a^b \sqrt{\left(\frac{dx}{dt}\right)^2 + \left(\frac{dy}{dt}\right)^2} dt \text{ for a parametric function}$$

*Note 1: The “matrix” section was removed from the syllabus a few years ago, but there is a chance it will come back in the next years (2024~25 or so).

*Note 2: I did not add much detail for obvious contents such as for exponential and logarithmic functions and calculus.

Japanese

1. Section A
 - a. JLPT N5 ~ N4
2. Section B
 - a. JLPT N3 ~ N2
3. Section C
 - a. JLPT N2 ~ N1
 - b. Additional proverbs and expressions not covered in JLPT

JLPT N5 →	30/300 ≈ 10%
JLPT N4 →	60/300 ≈ 20%
JLPT N3 →	120/300 ≈ 40%
JLPT N2 →	190/300 ≈ 60%
JLPT N1 (100/180 ~ 130/180) →	220/300 ≈ 70%
JLPT N1 (130/180 ~ 150/180) →	220/300 ≈ 80%
JLPT N1 (160/180+) →	270/300 ≈ 90%+

(Including lucky guesses)

Chemistry

*Note: Numbers or formulas with a ♦ sign must be memorized.

1. Constituent particles of matter
 - a. Simple substances
 - b. Compounds
 - c. Allotropes
 - ① SCOP (sulfur, carbon, oxygen, phosphorous)
 - d. (First) ionization energy
2. Chemical formulas
 - a. Molecular weight / formula weight
 - b. Avogadro's number
 - ① $6.02 \cdot 10^{23}$ ♦
 - c. Standard state
 - ① The volume of any (ideal) gas is 22.4L ♦ at standard state, and standard state always means 101 kPa at 0°C ♦
3. Chemical bonds and crystals
 - a. Ion, covalent, metallic bonds and intermolecular bonds (van der Waals force, hydrogen bond)
 - b. Electronegativity
 - c. Molecular polarity
 - d. NaCl, CsCl, diamond, body-centered cubic lattice, face-centered cubic lattice, and hexagonal close-packed structure
 - ① Number of atoms per unit cube, coordination number, and atomic packing factor ♦
4. States of matter
 - a. Vapor pressure
 - b. Vapor-liquid equilibrium
 - c. Ideal gas law
5. Solution
 - a. Henry's law
 - b. Concentration
 - ① Mass percent concentration (%)
 - ② Mole-concentration (mol/L)
 - ③ Molar concentration (mol/kg)
 - c. Osmotic pressure
 - ① $\Pi = cRT$
 - d. Colloids
 - ① Size of colloid particles: $10^{-9} \sim 10^{-7}\text{ m}$ ♦
 - ② Tyndall effect
 - ③ Brownian movement
 - ④ Electrophoresis
 - e. Hydrophobic and hydrophilic colloids
 - ① Coagulation
 - ② Salting out
 - ③ Dialysis
 - ④ Sol, gel
6. Chemical reactions and energy
 - a. Enthalpy
 - ① Enthalpy of formation
 - ② Enthalpy of dissolution
 - ③ Enthalpy of neutralization
 - b. Binding energy
 - c. Hess's law
7. Reaction speed
 - a. Activation energy
 - b. Catalysis
 - c. Chemical equilibrium
 - d. Le Chatelier's principle
 - e. Equilibrium constant
8. Acid and base
 - a. Arrhenius's theory
 - b. Bronsted-Lowry's theory
 - c. pH

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- d. Neutralization
 - ① Phenolphthalein ($\text{pH} \geq 7$)
 - ② Methyl orange ($\text{pH} \leq 7$)
- 9. Oxidation and reduction
 - a. Oxidizer, reducing agent
 - b. Battery
 - ① Primary, secondary
 - ② Voltaic battery
 $-\text{Zn}|\text{H}_2\text{SO}_4\text{aq}|{\text{Cu}} +$, reaction on each side ✎
 - ③ Daniell cell
 $-\text{Zn}|\text{ZnSO}_4\text{aq}|{\text{CuSO}_4\text{aq}}|{\text{Cu}} +$, reaction on each side ✎
 - ④ Manganese battery
 - ⑤ Lead-acid battery
 $-\text{Pb}|\text{H}_2\text{SO}_4\text{aq}|{\text{PbO}_2} +$, reaction on each side ✎
 - c. Ionization tendency
 - ① From bigger to smaller, Li K Ca Na Mg Al Zn Fe Ni Sn Pn (H₂) Cu Hg Ag Pt Au ✎
 - d. Electrolysis
 - e. Faraday's law
 - ① 96500 C/mol
- 10. Periodic table
 - a. Elements in group 1,2, 13 ~ 18 must be memorized in order (top to down) ✎
 - b. Transition elements and their characteristics
 - c. Main group elements and their characteristics
- 11. Non-metals
 - a. Halogens
 - b. Contact process ✎
 - c. Haber bosch process ✎
 - d. Ostwald process ✎
 - e. Solvay or ammonia-soda process ✎
- f. Oxoacid
- 12. Metals
 - a. Alkali metals
 - b. Alkaline earth metals
 - ① In Japan, only Ca, Sr, Ba, and Ra are considered alkaline earth metals due to their distinctive characteristics compared to Be and Mg. ✎
 - c. Aluminum
 - ① Production of aluminum, bauxite, aluminum oxyde
 - d. Hydroxide
 - e. Complex ions
 - ① $[\text{Ag}(\text{NH}_3)_2]^+$, straight line ✎
 - ② $[\text{Cu}(\text{NH}_3)_4]^{2+}$, square ✎
 - ③ $[\text{Zn}(\text{NH}_3)_4]^{2+}$, tetrahedron ✎
 - ④ $[\text{Fe}(\text{CN})_6]^{3-}$, octahedron ✎
- 13. Inorganic substances
 - a. Storage methods
 - ① Na, K, P, CaO, CaC₂, NaOH, AgNO₃, HF
 - b. The production method, collecting method, color, odor, acidity, solubility, and other characteristic features of the following gases must be memorized
 - ① H₂, O₂, Cl₂, HCl, HF, H₂S, SO₂, NO₂, NO, CO₂, CO, NH₃ ✎
 - c. Precipitation
- 14. Aliphatic compounds
 - a. Hydrocarbons
 - ① Alkane, cycloalkane
 - ② Alkene, cycloalkene
 - ③ Alkyne
 - b. The formula, naming, hydrolysis, acidity, and other characteristics of the following functional groups

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- must be memorized
- ① $-\text{OH}$, $-\text{O}-$, $-\text{CHO}$, =
 CO , $-\text{COOH}$, $-\text{COO}-$
 $, -\text{NH}_2$, $-\text{NO}_2$, $-\text{SO}_3\text{H}$ ♣
- c. Isomers
- ① Geometric, optical,
stereoisomer
- d. Alcohol
- ① Primary, secondary, tertiary
② Monohydric, dihydric, trihydric
- e. Ether
- f. Ester
- g. Carboxylic acid
- h. Oils and fats and soap
- ① Detergent
② Surface active agent
15. Aromatic compounds
- a. Phenol
- b. Aromatic carboxylic acid
- c. Aniline
- d. Pharmaceuticals
- ① Sulfonamides
② Antibiotics
- e. Dye, azo compounds
16. Natural polymers
- a. Sugar
- ① Monosaccharide
② Polysaccharide
- b. Amino acid
- ① Zwitterions
② Ninhydrin reaction
- c. Protein
- ① Peptide bond
② Biuret test
③ Xanthoprotein reaction
④ Enzyme
- d. Nucleic acid
- ① DNA, RNA
- ② Ribose, deoxyribose
17. Artificial polymers
- a. Fibers
- ① Rayon
② Acetate
③ Nylon (polyamide)
④ Vynylon
⑤ Acrylic fiber
⑥ Polyester
- b. Synthetic resin (plastic)
- ① Polyethylene
② Polyvinyl acetate
③ Phenolic resin
④ Urea resin
⑤ Polyethylene terephthalate
- c. Rubber
- ① Natural rubber (latex)
② Synthetic rubber,
vulcanization

Physics

1. Constant acceleration motion
 - a. V-t graph
 - ① $v = v_0 + at$
 - ② $x = v_0 t + \frac{1}{2}at^2$
 - ③ $v^2 - v_0^2 = 2ax$
 - b. Potential energy
 - ① $\frac{1}{2}mv^2$
 - ② Mgh
 - c. Relative velocity
2. Equilibrium of forces
 - a. Action-reaction law
 - b. Friction
 - c. Forces
 - ① mg
 - ② kx
 - ③ ρVg
 - d. Spring

$$\textcircled{1} \quad \frac{1}{k} = \frac{1}{k_1} + \frac{1}{k_2}, \quad k = k_1 + k_2$$
 - e. Center of mass

$$\textcircled{1} \quad x_G = \frac{m_1 x_1 + m_2 x_2 + m_3 x_3 + \dots}{m_1 + m_2 + m_3 + \dots}$$
3. Laws of motion
 - a. Newtonian equation of motion

$$\textcircled{1} \quad \vec{F} = m\vec{a}$$
 - b. Conservation of energy and work

$$\textcircled{1} \quad U = \frac{1}{2}kx^2$$

$$\textcircled{2} \quad W = Fl$$
4. Movements with resistance
 - a. Friction

$$\textcircled{1} \quad \mu'N$$
 - b. Air resistance

$$\textcircled{1} \quad kv$$
5. Conservation of momentum
 - a. Momentum and impulse

$$\textcircled{1} \quad m\vec{v}$$

- $\textcircled{2} \quad \vec{F} \cdot \Delta t$
- b. Coefficient of restitution and collision

$$\textcircled{1} \quad V' - v' = -e(V - v)$$
6. Circular motion and universal gravitation
 - a. Circular motion

$$\textcircled{1} \quad T = \frac{2\pi}{v} = \frac{2\pi}{\omega}$$

$$\textcircled{2} \quad a = \frac{v^2}{r} = r\omega^2$$

$$\textcircled{3} \quad F = m\frac{v^2}{r} = mr\omega^2$$
 - b. Universal gravitation

$$\textcircled{1} \quad f = G \frac{Mm}{r^2}$$

$$\textcircled{2} \quad U = -G \frac{Mm}{r}$$
 - c. Kepler's law

$$\textcircled{1} \quad \frac{1}{2}rv = \text{constant}$$

$$\textcircled{2} \quad T^2 = ka^3$$
 - d. Escape velocity

$$\textcircled{1} \quad \frac{1}{2}mv_0^2 + \left(-G \frac{Mm}{r}\right) \geq 0$$
7. Harmonic motion and pendulum
 - a. Simple harmonic motion

$$\textcircled{1} \quad x = A \sin \omega t$$

$$\textcircled{2} \quad v = A\omega \cos \omega t, \quad v_{\max} = A\omega$$

$$\textcircled{3} \quad T = \frac{2\pi}{\omega}$$
 - b. Spring pendulum

$$\textcircled{1} \quad T = 2\pi \sqrt{\frac{m}{k}}$$
 - c. Pendulum

$$\textcircled{1} \quad T = 2\pi \sqrt{\frac{l}{g}}$$

$$\textcircled{2} \quad g' = g + a \text{ in an elevator}$$

$$\textcircled{3} \quad g' = \sqrt{g^2 + a^2} \text{ in a car}$$
8. Temperature and heat
 - a. Heat capacity

$$\textcircled{1} \quad Q = mc\Delta t$$

$$\textcircled{2} \quad Q = C\Delta t$$

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- b. Thermal efficiency
- $$\textcircled{1} \quad e = \frac{W}{Q_{\text{in}}} = \frac{Q_{\text{in}} - Q_{\text{out}}}{Q_{\text{in}}} < 1$$
9. Motion of particles and change in state
- a. Equation of state and laws of gas
- $$\textcircled{1} \quad PV = nRT$$
- $$\textcircled{2} \quad pS = p_0S + Mg$$
- b. Movement of particles
- $$\textcircled{1} \quad \frac{1}{2}m\bar{v^2} = \frac{3R}{2N_A}T = \frac{3}{2}nRT$$
- $$\textcircled{2} \quad \Delta U = nC_V\Delta T$$
- $$\textcircled{3} \quad W = p\Delta V = nR\Delta T$$
- c. Change in state
- $$\textcircled{1} \quad C_V = \frac{3}{2}R, \quad C_P = C_V + R$$
- $$\textcircled{2} \quad pV^\gamma = \text{constant}, \quad \gamma = \frac{C_P}{C_V}$$
- $$\textcircled{3} \quad Q = \Delta U + W$$
10. Properties of waves
- a. Y-t and y-x graphs
- $$\textcircled{1} \quad y = A \sin 2\pi \left(\frac{t}{T} - \frac{x}{\lambda} \right)$$
- b. Fixed end and free end reflection
- c. Interference
- $$\textcircled{1} \quad |l_1 - l_2| = m\lambda \text{ or } = \left(\frac{1}{2} + m \right) \lambda$$
- d. Sound waves
- $$\textcircled{1} \quad \text{String: } v = \sqrt{\frac{s}{\rho}}, \quad f_m = \frac{m}{2l} \sqrt{\frac{s}{\rho}} \quad (m = 1, 2, 3, \dots)$$
- $$\textcircled{2} \quad \text{Open pipe: } f_m = \frac{mV}{2l} \quad (m = 1, 2, 3, \dots)$$
- $$\textcircled{3} \quad \text{Closed pipe: } f_m = \frac{mV}{4l} \quad (m = 1, 3, 5, \dots)$$
- e. Doppler effect
- $$\textcircled{1} \quad f' = \frac{V-u}{V+u} f_0$$
- f. Beat
11. Light waves
- a. Properties of light
- $$\textcircled{1} \quad n = \frac{c}{v}$$
- $$\textcircled{2} \quad n_{12} = \frac{v_1}{v_2} = \frac{\lambda_1}{\lambda_2} = \frac{\sin \theta_1}{\sin \theta_2} = \frac{n_2}{n_1}$$
- $$\textcircled{3} \quad n_1 \sin \theta_1 = n_2 \sin \theta_2$$
- b. Lenses
- $$\textcircled{1} \quad \frac{1}{f} = \frac{1}{a} + \frac{1}{b}$$
- $$\textcircled{2} \quad m = \left| \frac{b}{a} \right|$$
- c. Young's experiment
- $$\textcircled{1} \quad \lambda m = \frac{dx}{l}$$
- d. Newton's rings
- $$\textcircled{1} \quad d = \frac{r^2}{2R}$$
12. Electrostatic force and electric field
- a. Formulas and laws
- $$\textcircled{1} \quad F = k \frac{Qq}{r^2}$$
- $$\textcircled{2} \quad E = k \frac{Q}{r^2}$$
- $$\textcircled{3} \quad V = k \frac{Q}{|r|}$$
- $$\textcircled{4} \quad N = 4\pi kQ = \frac{Q}{\epsilon_0}$$
- $$\textcircled{5} \quad U = qV$$
13. Capacitor
- a. Capacitor laws
- $$\textcircled{1} \quad \text{Note: In Japan, capacitors are called "condensers"}$$
- $$\textcircled{2} \quad Q = CV$$
- $$\textcircled{3} \quad C = \epsilon \frac{s}{d}$$
- $$\textcircled{4} \quad U = \frac{1}{2} QV$$
- $$\textcircled{5} \quad F = \frac{CV^2}{2d}$$
- b. Conservation of electric power
14. DC circuit

MEXT Undergrad Syllabus

- a. Current laws
- ① $I = S_{\text{env}}$
 - ② $V = RI$
 - ③ $V = E - rl$
15. Electric current and magnetic field
- a. Magnetic field
- ① $H = \frac{I}{2\pi r}, H' = \frac{I}{2r}, H'' = nI$
 - ② $B = \mu H$
 - ③ $eE = evB$
 - ④ $E = vB$
16. Electromagnetic induction
- a. Induction laws
- ① $V = -N \frac{\Delta\phi}{\Delta t}$
 - ② $V = Blv_{\perp} = Blv \cos \theta$
- b. Self and mutual induction
- ① $V = -L \frac{\Delta I}{\Delta t}$
 - ② $V' = -M \frac{\Delta I}{\Delta t}$
 - ③ $U = \frac{1}{2}LI^2$
17. AC circuit
- a. Generator of AC
- ① $V = V_0 \sin \omega t$
 - ② $\varphi = BS \cos \omega t$
- b. AC circuit
- ① $V = RI$
 - ② $V = \omega LI$
 - ③ $V = \frac{1}{\omega C}I$
 - ④ $Z = \sqrt{R^2 + (\omega L - \frac{1}{\omega C})^2}$
- c. Resonance circuit
- ① $T = 2\pi\sqrt{LC}$
 - ② $\frac{1}{2}LI^2 = \frac{1}{2}CV^2$
18. Electrons and lights
- a. Electron in a magnetic field
- ① $F = eE$
- ② $\frac{1}{2}mv_0^2 + eV = \frac{1}{2}mv^2$
- ③ $a = \frac{eE}{m}$
- ④ $f = evB$
- b. Wave-particle duality of light
- ① $E = hv$
 - ② $hv = W + \frac{1}{2}mv^2$
 - ③ $\lambda = \frac{c}{v}$
- c. X-rays
- ① $\lambda = \frac{hc}{eV}$
- d. Compton effect
- e. Bragg's law
- ① $2d \sin \theta = n\lambda \ (n = 1,2,3,\dots)$
19. Atoms and nuclei
- a. Hydrogen atom
- ① $m \frac{v^2}{r} = k \frac{ee}{r^2}$
 - ② $2\pi r = n \frac{h}{mv} \ (n = 1,2,3,\dots)$
 - ③ $E = -\frac{ke^2}{2r}$
- b. Release of light
- ① Lyman series
 - ② Balmer series
 - ③ Paschen series
 - ④ $\frac{1}{\lambda} = R \left(\frac{1}{n'^2} - \frac{1}{n^2} \right)$
- c. Radioactive decay
- ① $\frac{N}{N_0} = \left(\frac{1}{2}\right)^t$
- d. Properties of atoms and energy
- ① $E = mc^2$